



Matter Meets Anti-Matter

Probing Polymers with Positrons

The University of Michigan
Positron Group

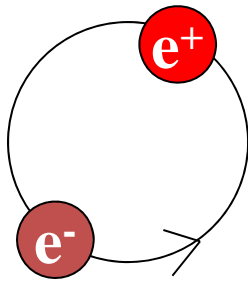
Professor David Gidley
Dr. Ming Liu
Daniel Blatter
Rob Davis



Why Positrons?

- Positron Annihilation Lifetime Spectroscopy (PALS) uses Positrons (e^+) as probes to investigate materials
- When injected, e^+ forms Positronium (Ps), finds vacancies in the material, and annihilates with a lifetime corresponding to the size of the vacancies.
- PALS is very sensitive to the “free volume” in a material

Positronium, Ps ($Z=1$, $A=0$)



Ps is its own anti-atom !

Hydrogen-like bound state

$\frac{1}{2}$ Rydberg (6.8 eV) binding energy

Spin singlet, $S=0$ (para-positronium): 1/8 ns lifetime, 2 γ rays

Spin triplet, $S=1$ (ortho-positronium): 142 ns lifetime, 3 γ rays

Positronium forms naturally in materials and ortho-Ps will be our probe



Radioactive ^{22}Na Decay

^{22}Na decays in (essentially) two ways

- Beta decay (90% of decays):



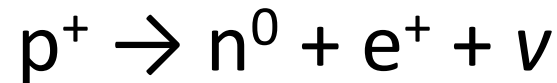
- Electron capture (10% of decays):





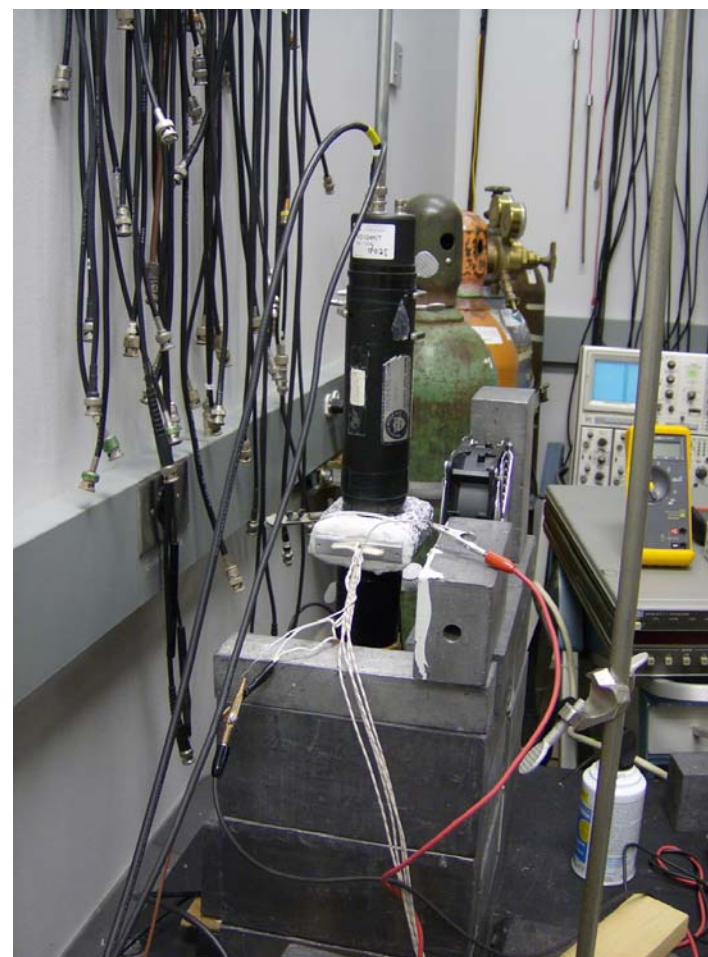
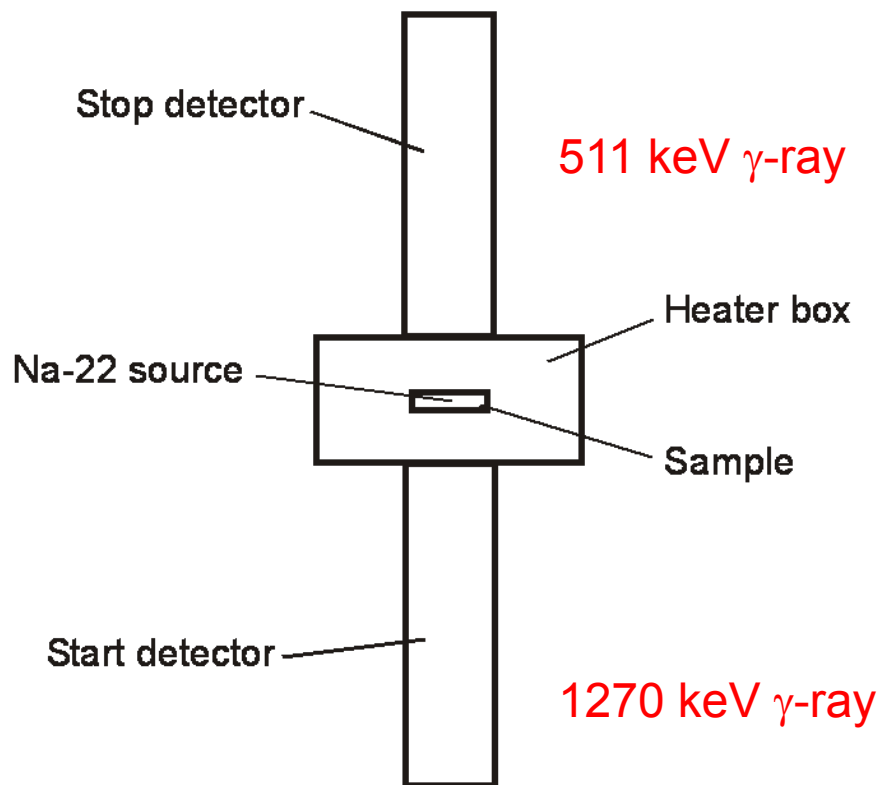
Beta Decay in ^{22}Na

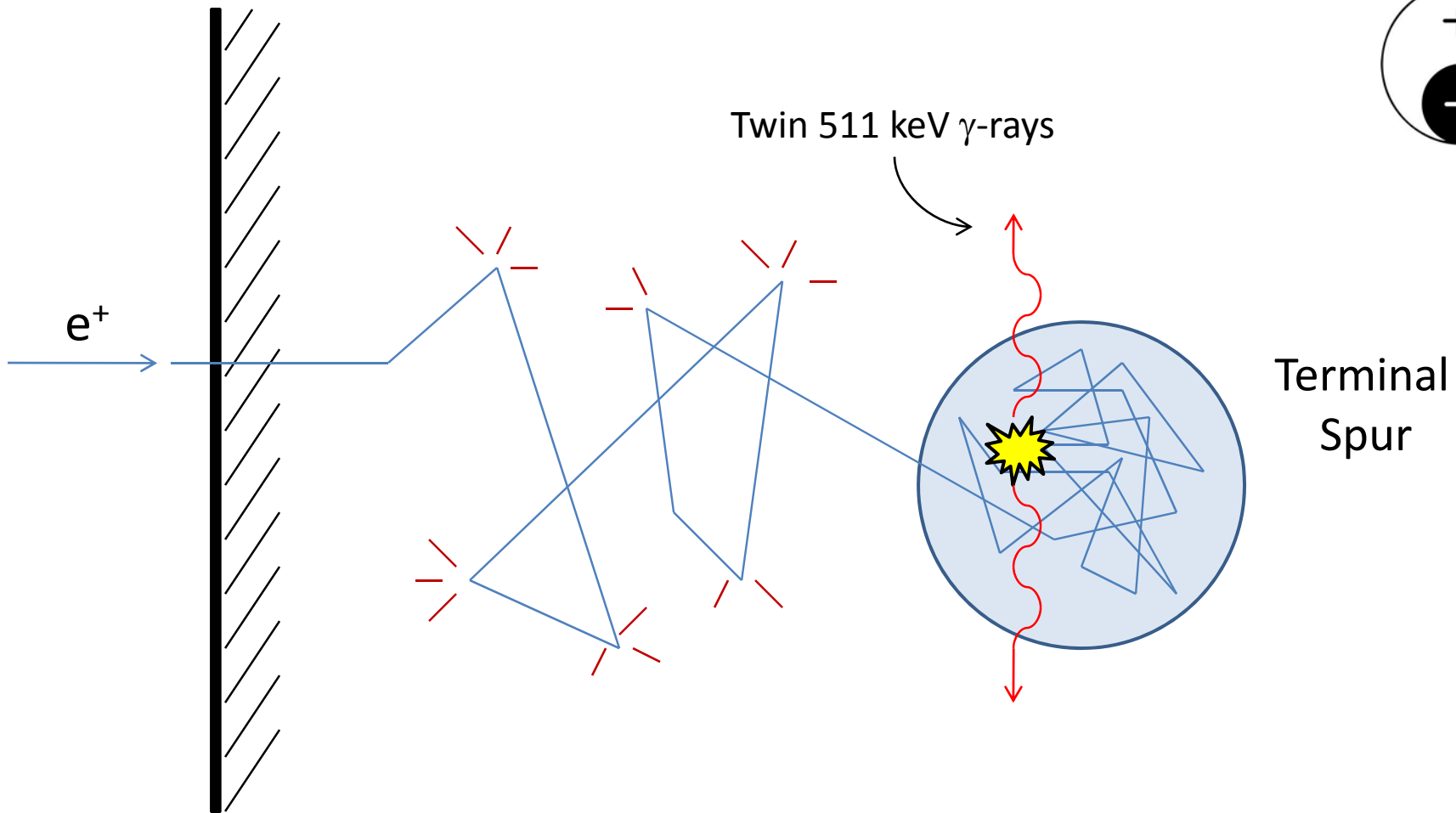
- β^+ decay involves the following process:



- The emitted e^+ have an energy distribution ranging from 0 - 0.54 MeV, with a peak in the distribution at 0.19 MeV
- Penetration of $\sim 1\text{mm}$ in polymers.

Bulk PALS (Positronium Annihilation Lifetime Spectroscopy)





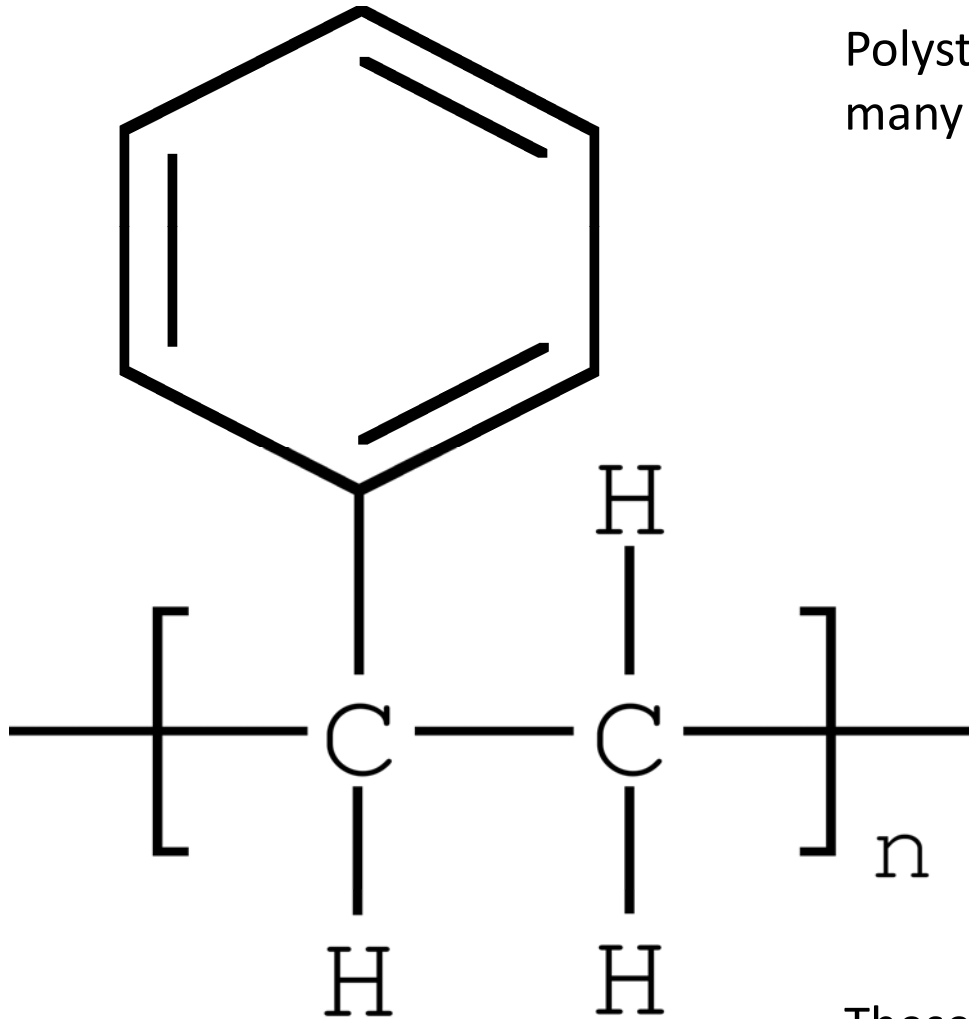
Positron behavior in a solid

The 'hot' ($\sim 200\text{keV}$) positrons bounce around, ionizing atoms in the material. With each collision, they lose energy. After $\sim 10^4$ collisions, they have thermalized ($\text{KE} \sim 1\text{eV}$) and can form Ps.



Polystyrene is made by linking many styrene subunits together.

One styrene subunit has a molecular weight (M_w) of 104.

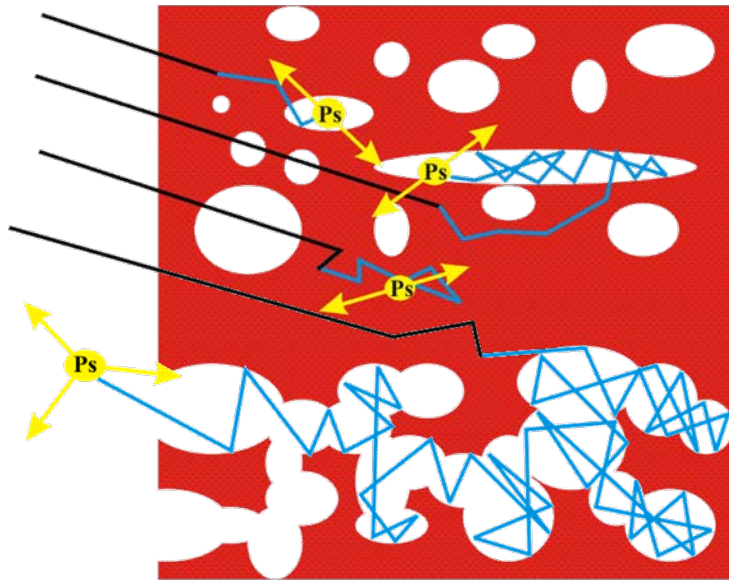


One Styrene Subunit

152k M_w PS, for example, is a chain of styrene groups approximately 1,460 units long!

These long chains fold around each other like a bowl of spaghetti noodles. The spaces left between noodles are the voids we probe using PALS!

Ps Formation in Nanoporous Materials

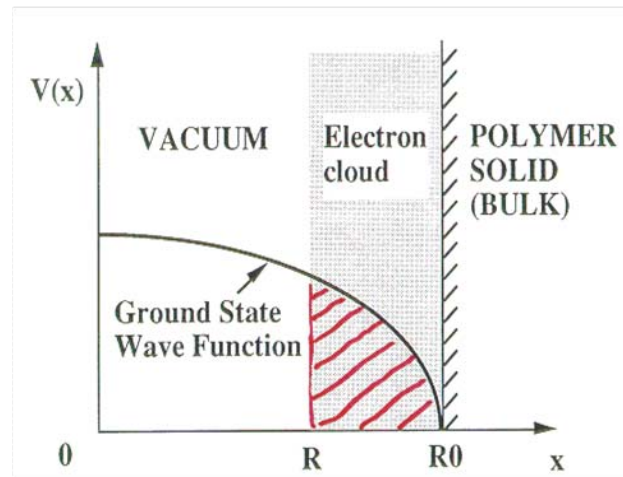
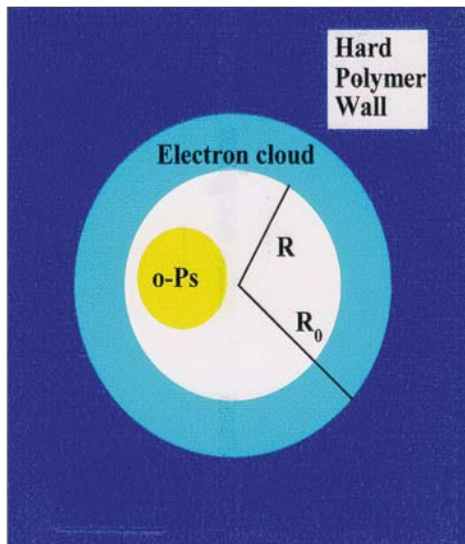


Thermalized e^+ in a nanoporous material either:

1. annihilates directly with e^-
2. forms para-Ps (lifetime ~ 125 ps)
3. forms ortho-Ps (lifetime ~ 142 ns)

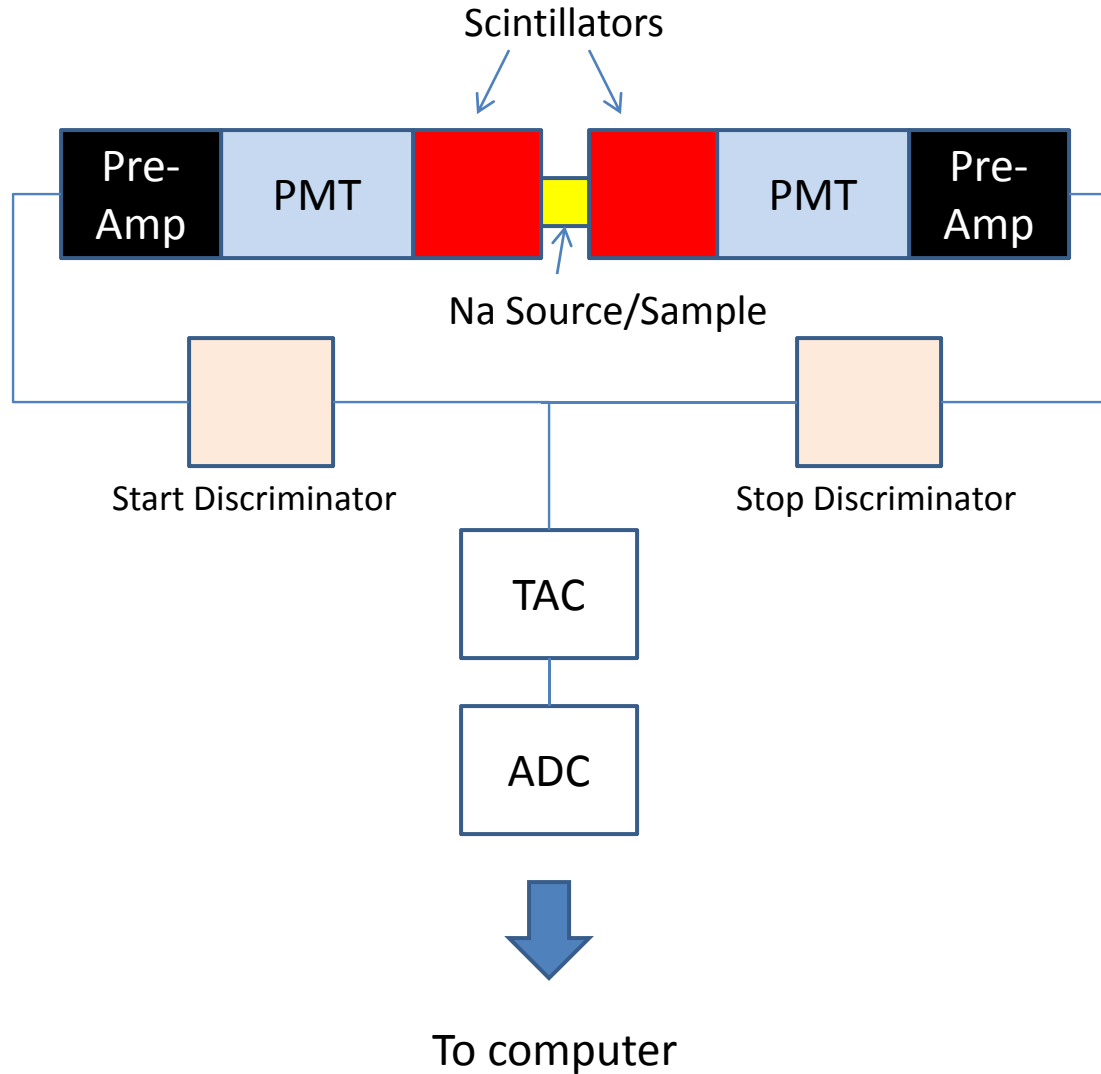
Ortho-Ps tends to find nanopores in the material (due to dielectric response of the medium)

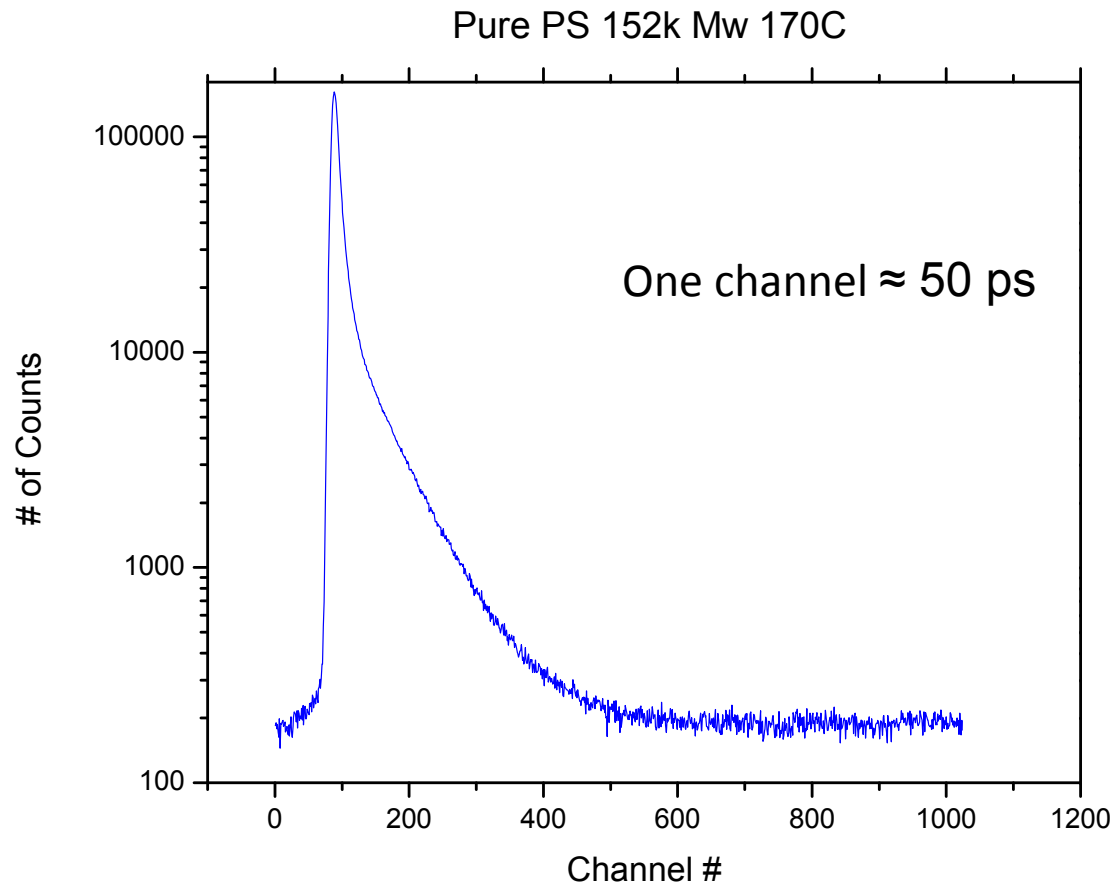
The lifetime of o-Ps is reduced by interaction with the electrons 'spilling' into the void from its Walls (from 142ns in vacuum to ~ 1 ns in voids)





Data Taking Apparatus





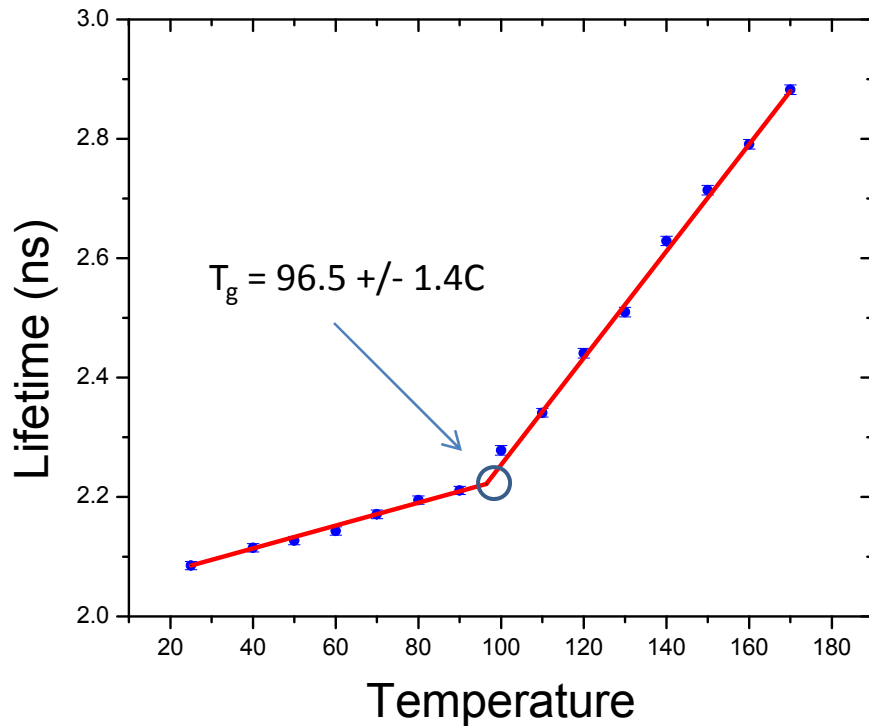
A typical spectrum

Time $t = 0$ corresponds to the 'prompt peak.' The offset is to allow the full spectrum to appear on the graph. The y-axis is a log-scale, so the straight portions of the spectrum indicate exponential decays. Each slope corresponds to a lifetime that we determine by fitting.

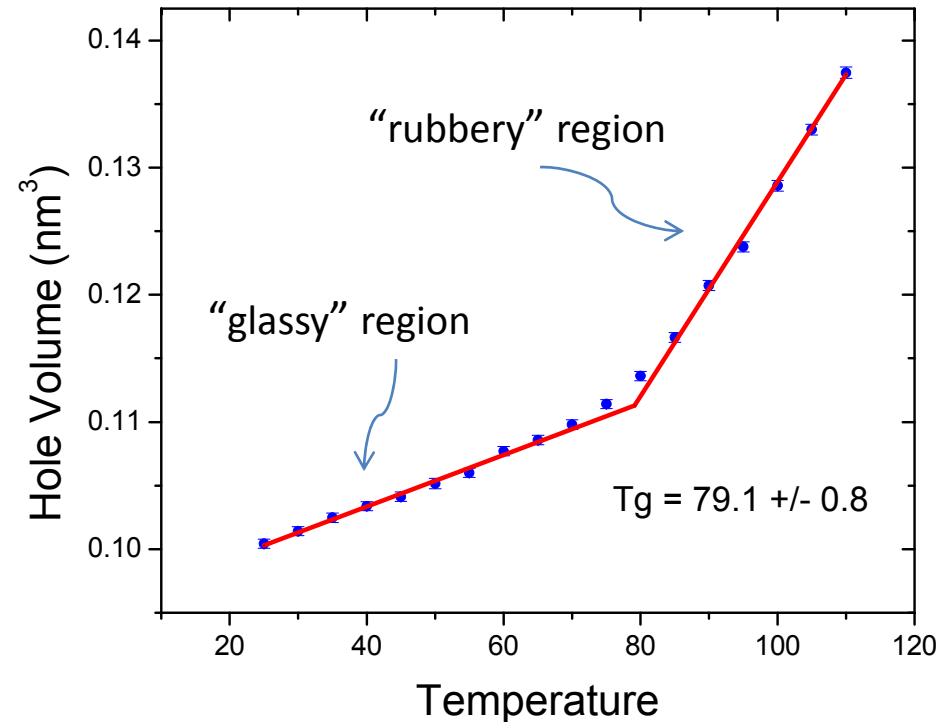


Glass Transition of a Polymer

Pure PS 152k scan 4



Pure PS 5.2k Average Hole Volume

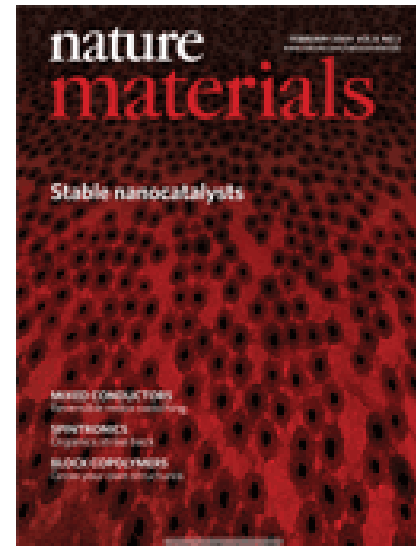


T_g is determined by finding the intersection of the fitted lines in each region



Materials Science Research

- Peter Green and Hyunjoon Oh reported in *Nature Materials* that the addition of gold nanoparticles (to which PS chains of varying lengths were grafted) to pure 5.2k M_w PS caused significant changes in glass transition.
- Grafted chains were 481 and 10 subunits long



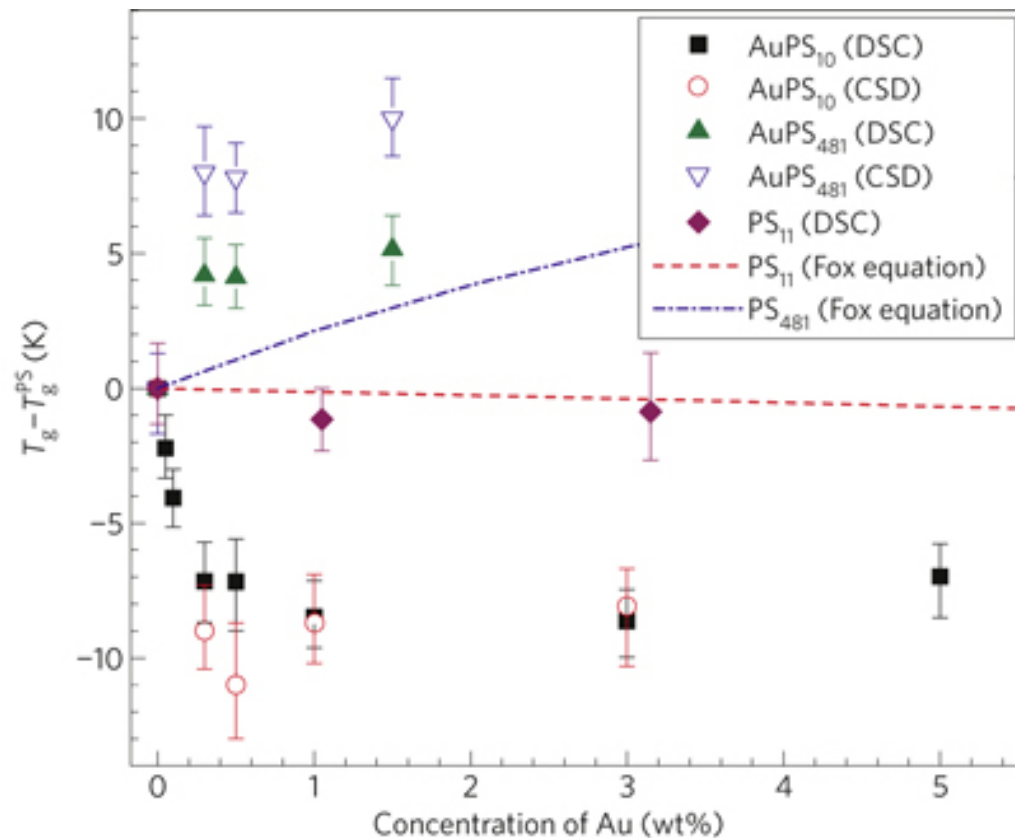
Green, P. Oh, H.

“Polymer chain dynamics and glass transition in athermal polymer/nanoparticle mixtures.”
Nature Materials 8, 139 – 143 (2009)



Gold Polymer Nanocomposites

- AuPS₁₀ nanoparticles increase the free volume, lowering T_g by ~ 7 °C
- AuPS₄₈₁ nanoparticles reduce free volume, raising T_g by ~ 5 °C
- Green et al. used DSC and CSD to measure T_g . These methods are less sensitive to free volume than PALS
- They asked our group to measure T_g using PALS



Green, P. Oh, H. "Polymer chain dynamics and glass transition in athermal polymer/nanoparticle mixtures." *Nature Materials* 8, 139 – 143 (2009)

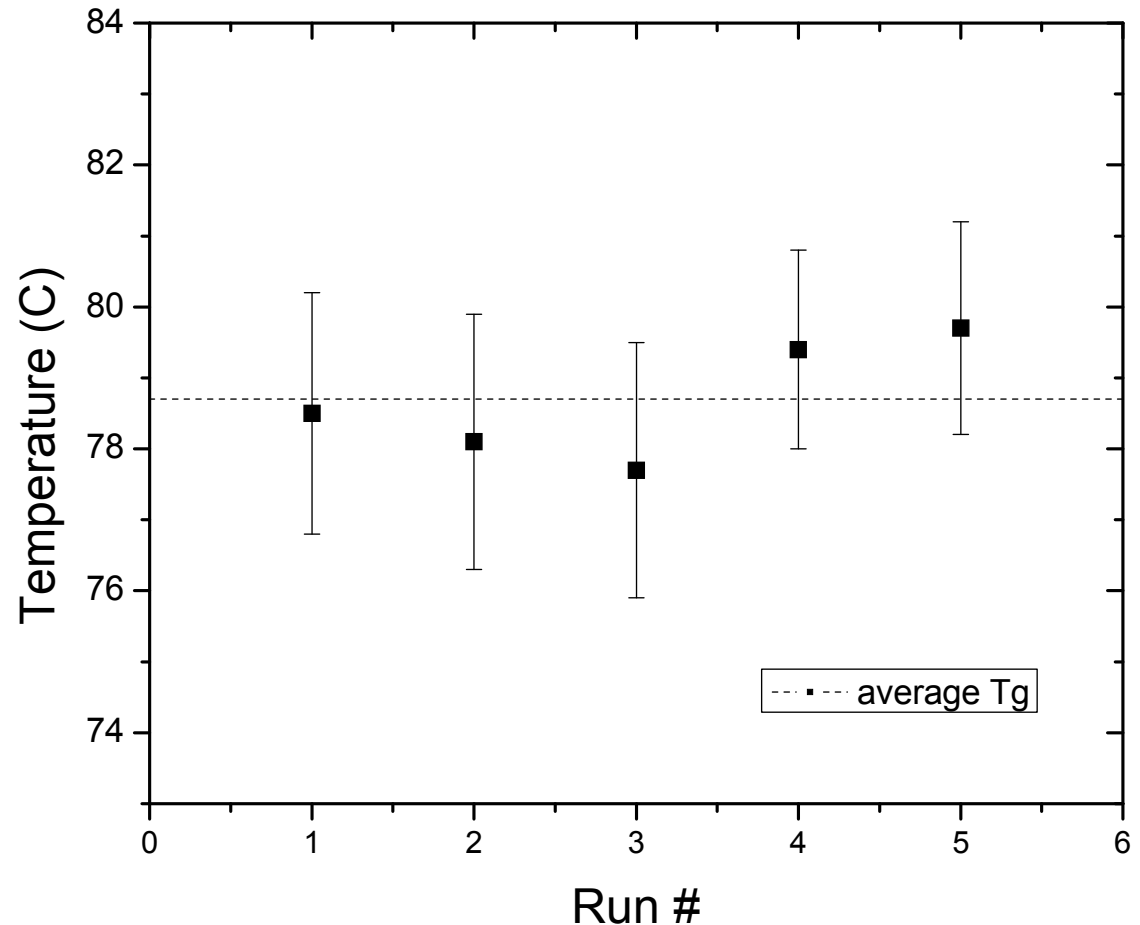


Pure Polystyrene Matrix

- The matrix into which the AuPS₁₀ and AuPS₄₈₁ nanoparticles were added is pure 5.2k PS (N = 50).
- We analyzed pure 5.2k PS using PALS and deduced a glass transition of 78.7

$$T_g = 78.7 \pm 0.8 \text{ } ^\circ\text{C}$$

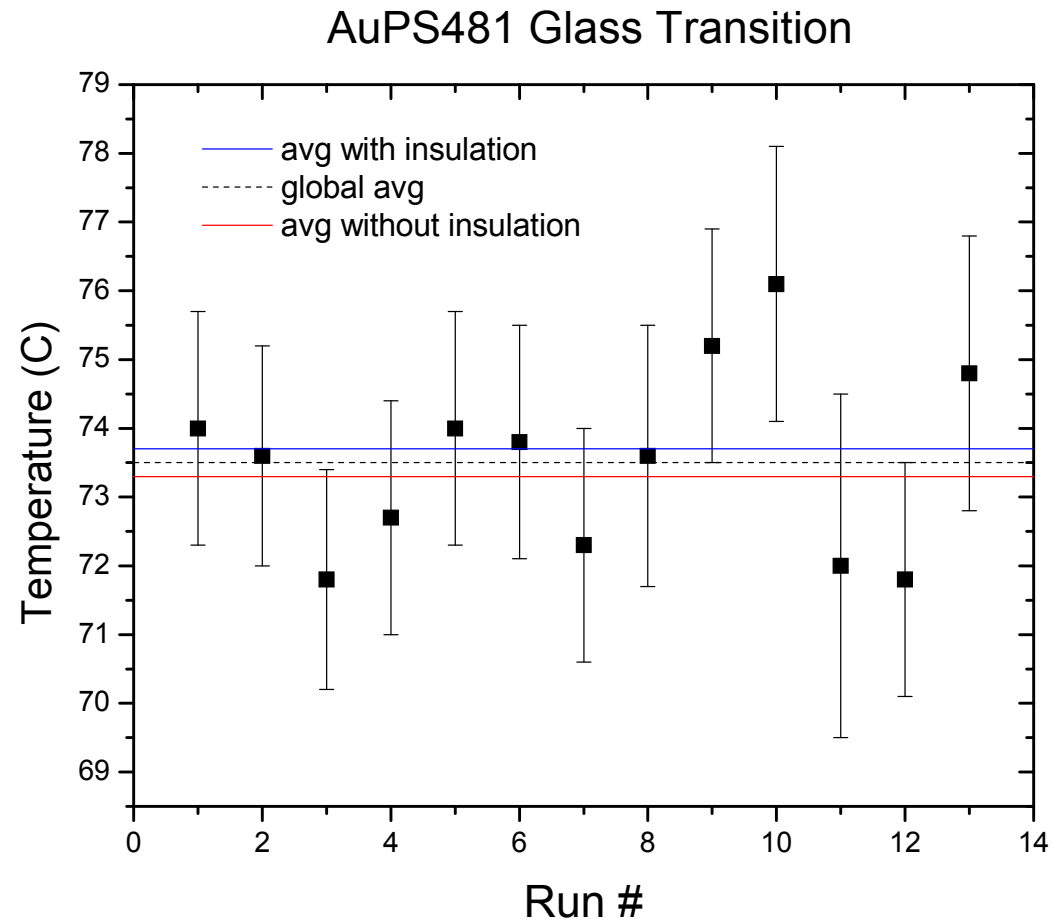
Pure PS 5.2k Glass Transition Temperature



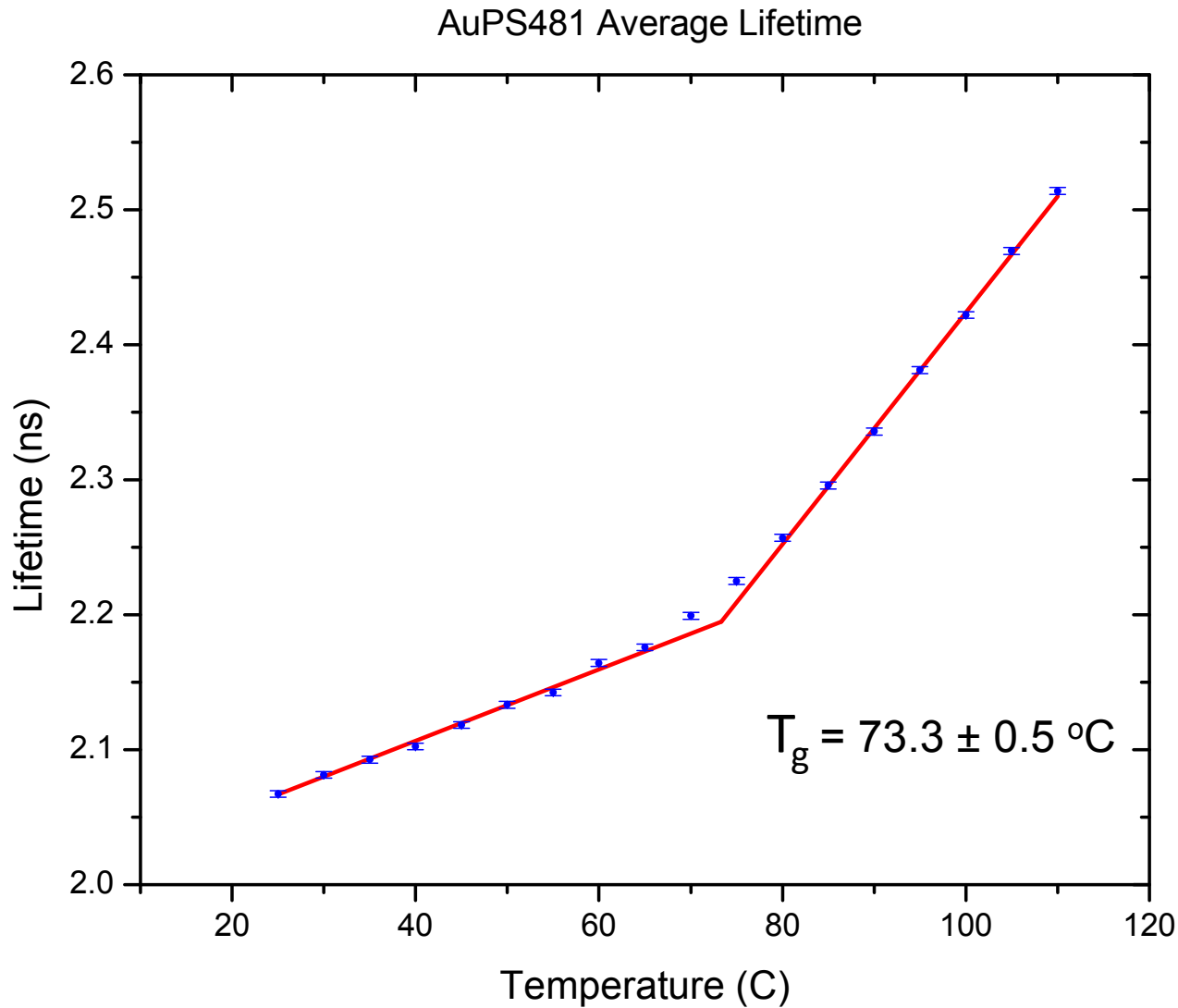


Our Results – AuPS₄₈₁ Glass Transition

- We deduced an T_g of 73.5 ± 0.5 °C. This represents a ΔT_g of ~ -5 °C.
- For AuPS₄₈₁ Peter Green et al. predicted a ΔT_g of $\sim +5$ °C!



Fitted T_g Using Averaged Lifetime

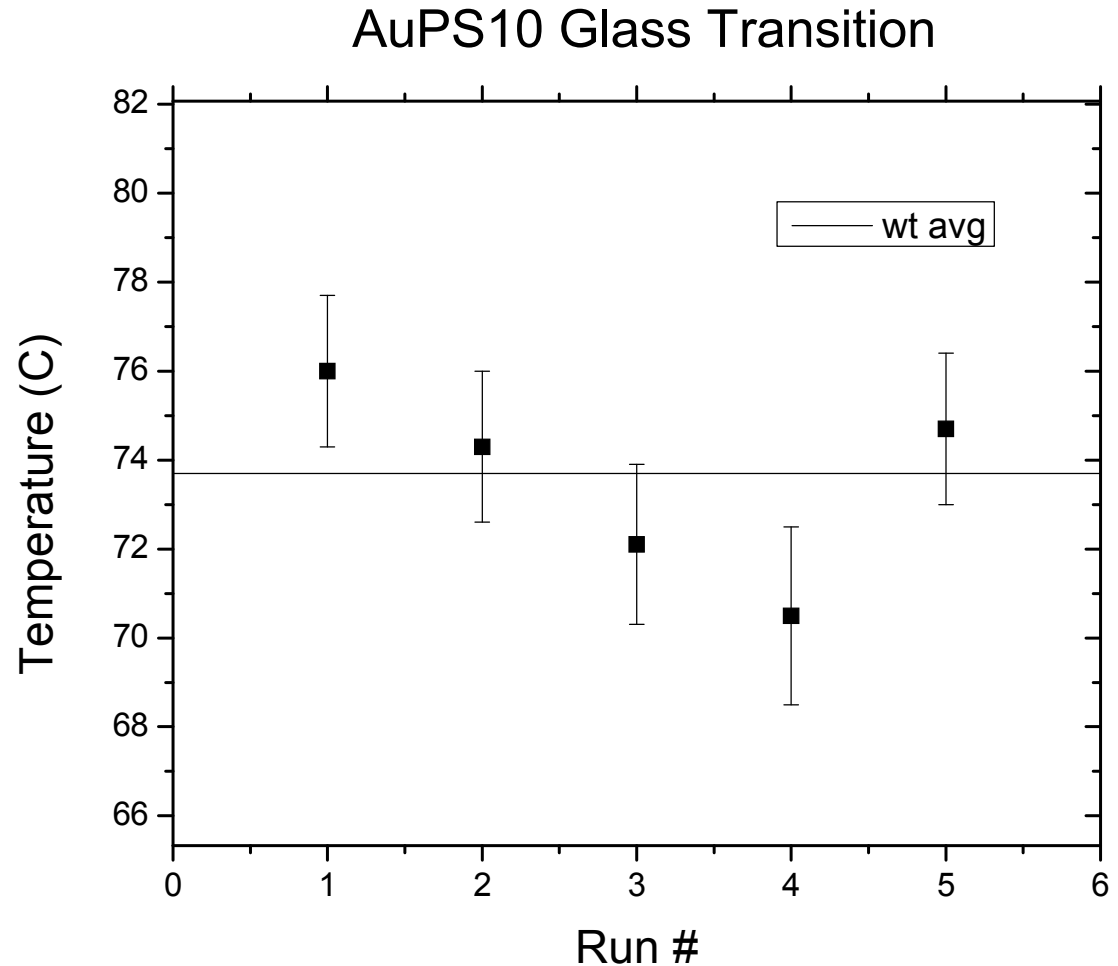


Our Results – AuPS₁₀ Glass Transition



- We analyzed pure 5.2k PS matrix (chains with 50 subunits) with gold nanoparticles whose attached PS chains were only 10 subunits long.
- We deduced a glass transition of 73.7 ± 0.8 °C.

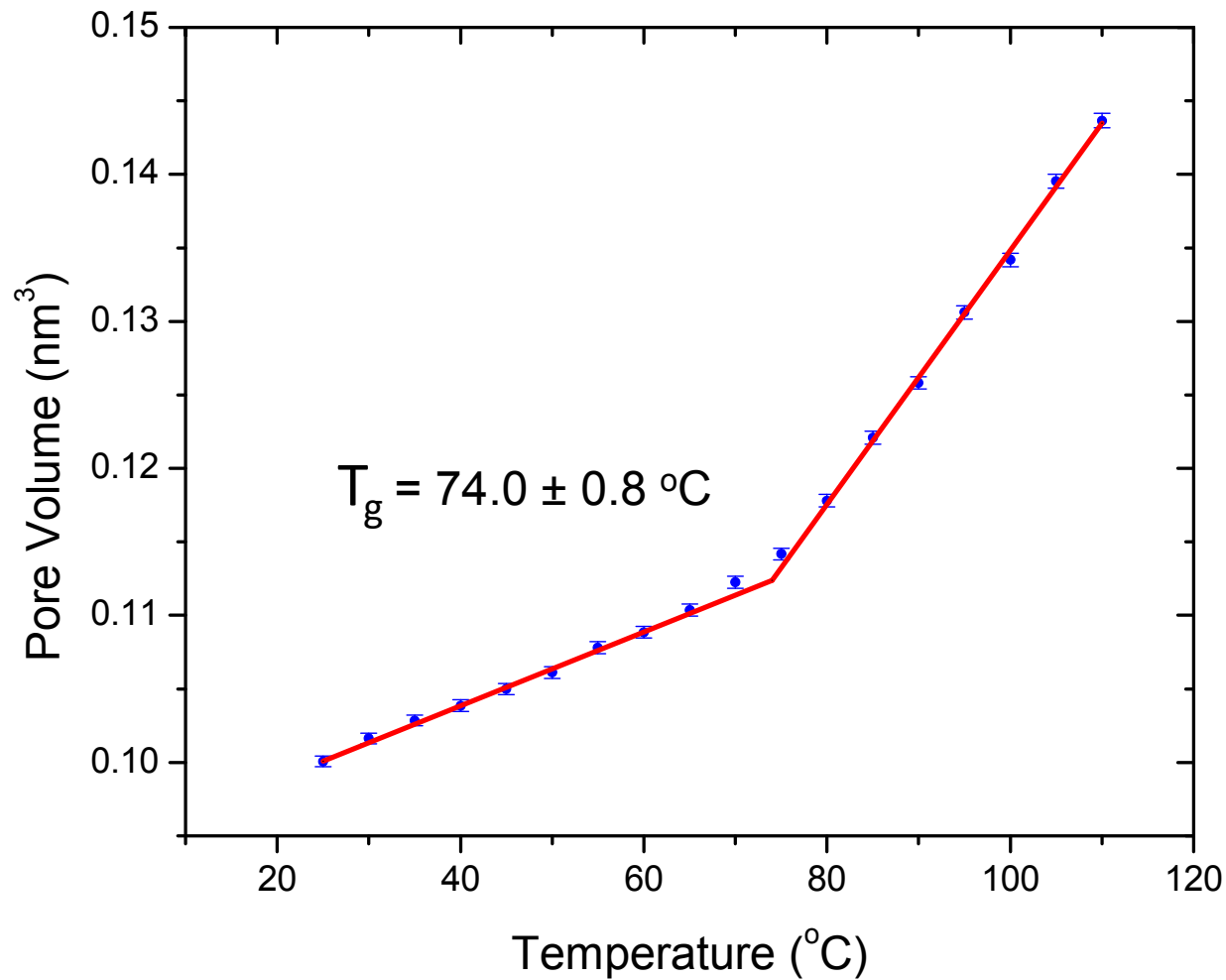
$$T_g = 73.7 \pm 0.8 \text{ °C}$$



Fitted T_g Using Pore Volume



AuPS10 Pore Volume





Conclusion: ΔT_g

AuPS₁₀

- For AuPS₁₀, we measured:
 $\Delta T_g \approx - 5 \text{ }^\circ\text{C}$.
- Our colleagues in Materials Science reported, using DCS and CSD: $\Delta T_g \approx - 7 \text{ }^\circ\text{C}$
- Our research confirms their result

AuPS₄₈₁

- For AuPS₄₈₁, we measured:
 $\Delta T_g \approx - 5 \text{ }^\circ\text{C}$.
- Our colleagues reported:
 $\Delta T_g \approx + 5 \text{ }^\circ\text{C}$ - a discrepancy of 10 °C!
- We are currently confirming this result with a new AuPS₄₈₁ sample.
- Results should be in by Friday!

Acknowledgements

I would like to sincerely thank

Professor David Gidley

&

Dr. Ming Liu

Also many thanks to:

Rob Davis – University of Michigan positron group

Professor Peter Green – Materials Science and Engineering

Dr. Emmanouil Glynos – Materials Science and Engineering

Paul Dirac – for predicting the existence of the positron

This summer REU was made possible by the NSF, the University of Michigan Physics Department, and the hard work of Professor Jim Liu and others. Many thanks!



Arthur Rich Memorial
University of Michigan
Sculpture/photo: Jens Zorn